

FIGURE 4.-GEOLOGIC SECTIONS.

INTRODUCTION In many parts of the Fort Peck Indian Reservation, which is located in northeastern Montana (fig. 1), ground water is the only source of water supply. Recently, some tribal residents have become concerned that the quality of water from their domestic wells has deteriorated. In response to this concern, the U.S. Geological Survey, in cooperation with the Water Resources Office of the Fort Peck Tribes, made a reconnaissance study of the ground-water resources of

Purpose and Scope

the reservation.

This report describes the results of the reconnaissance study, which was conducted during 1989-90. Specifically, the report describes the generalized geology and ground-water resources, and presents the well-inventory and water-quality data collected during 1989. As part of the study, existing geologic and hydrologic data were compiled, additional wells and springs were inventoried, short-term water-level fluctuations in wells were monitored, water samples were collected from selected wells and analyzed for chemical quality, and all information was put in a com-

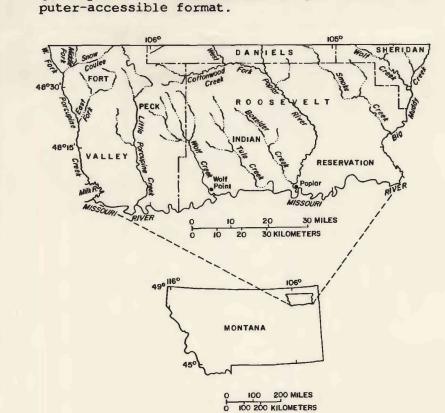


FIGURE 1.-LOCATION OF STUDY AREA.

General Description of Area The Fort Peck Indian Reservation includes an area of about 3,280 mi² in parts of Daniels, Roosevelt, Sheridan, and Valley Counties. The reservation is bounded by Porcupine Creek and the

Milk River on the west, Big Muddy Creek on the

east, and the Missouri River on the south. On

the north, the border is the 48°38' parallel,

which is 40 mi north of the mouth of the Milk

Three general types of terrain characterize the area. The Missouri River valley bottom is irrigated farmland with deciduous trees. Near the major Missouri River tributaries, which dissect the reservation, the land surface is generally rolling to rough rangeland (badlands) used primarily for livestock grazing. High benchlands, the third type of terrain, are characterized by relatively fertile soils, which are well suited for the production of small grains.

The reservation has a semiarid climate that is typical of the northern Great Plains. Normal annual precipitation ranges from about 12.0 to 13.5 in., with most precipitation received in June. Mean annual temperature ranges from about 40 to 52 °F. The mean monthly temperatures range from 7.1 °F in January to about 71 °F in July

(National Oceanic and Atmospheric Administration,

Well and Spring Numbering System In this report, wells and springs are numbered according to geographic position within the rectangular grid system used in Montana by the U.S. Bureau of Land Management. The location (local) number consists of 14 characters. The first two characters specify the township and its position north (N) of the Montana Base Line. The next two characters specify the range and its position east (E) of the Montana Principal Meridian. The next two characters are the section number. The next four characters designate the quarter section (160-acre tract), quarter-quarter section (40-acre tract), quarter-quarter-quarter section (10-acre tract), and quarter-quarterquarter-quarter section (2.5-acre tract), respectively, in which the well or spring is located. The subdivisions of the section are designated A, B, C, and D in a counterclockwise direction, beginning in the northeast quadrant. The last two characters form a sequence number indicating the order of inventory. For example, as shown in figure 2, well 31N52E05CDDB02 is the second well inventoried in the NW1/4 of the SE1/4 of the

SE1/4 of the SW1/4 of section 5, in Township 31

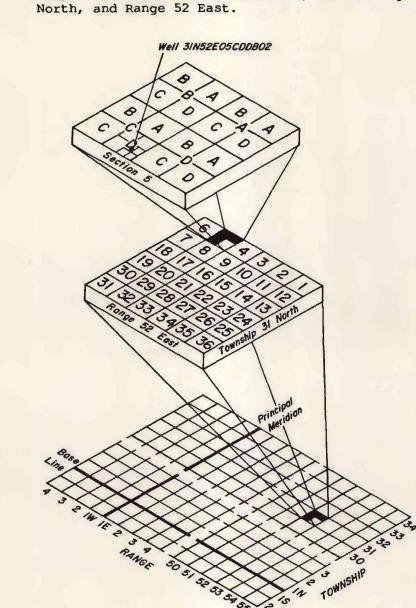


FIGURE 2 .-- WELL AND SPRING NUMBERING SYSTEM

Data Collection As an initial step in establishing a monitoring-well network, 118 wells and 2 springs (fig. 3) were inventoried to supplement existing geologic and hydrologic data. The data collected onsite include location, type of site, altitude of land surface, use of water, depth of well, water level, water-level status, altitude of water level, specific conductance, pH, water temperature, and principal aquifer (table 1). Available lithologic well logs were obtained for

hydrologic analysis. A monitoring network of 53 wells was selected using inventory data. The wells were selected to represent geographically the major aquifers in the reservation. Water levels were measured twice during 1989 at each well in the network; specific conductance, pH, and water temperature were also measured twice, where possible.

Water samples for chemical analysis were collected from 20 selected wells during July 1989. Water samples were measured for physical properties and analyzed for major-ion concentrations (table 2) and trace-element and organic-carbon concentrations (table 3). In addition, nine water samples were analyzed for the stable isotopes of oxygen and hydrogen (table 3). The water samples were analyzed by the U.S. Geological Survey National Water Quality Laboratory in Denver, Colo.

Acknowledgments

Appreciation is expressed to the many landowners who permitted access to their property and provided information about their wells and springs. Appreciation is also extended to the Water Resources Office and the Office of Environmental Protection of the Fort Peck Tribes and to the Montana Bureau of Mines and Geology for supplying data for existing wells.

GENERALIZED GEOLOGY

The generalized geology is shown in figure 3; glacial deposits thinner than about 80 ft, which cover the bedrock units in most of the reservation, are not shown on the map. Except for the Bearpaw Shale, the mapped geologic units, which range in age from Late Cretaceous to Holocene, yield most of the ground water used in the reservation. The generalized thickness and structure are shown in the geologic sections (fig. 4).

The Upper Cretaceous Bearpaw Shale consists of dark clayey shale. This unit underlies the entire reservation and is exposed at land surface in southern and western areas. Where exposed, the shale forms either badlands or small hills having gentle slopes (Jensen and Varnes, 1964). The Upper Cretaceous Fox Hills Sandstone consists of an upper sandstone unit and a lower marine shale. This formation is exposed at land surface in areas east of Little Porcupine Creek and north of the Missouri River near Wolf Point. The Upper Cretaceous Hell Creek Formation

are capped by sandstone ledges. The Paleocene Fort Union Formation consists dominantly of gray clay, buff silt, buff sandstone, and olive-gray sand irregularly interbedded with dark carbonaceous seams and thin and discontinuous lignite coal beds (Witkind, 1959). The Fort Union Formation is exposed at land surface in areas between Little Porcupine Creek and Smoke Creek in the north-central part of the res-

consists dominantly of shale, siltstone, and

sandstone. The formation is exposed at land

surface in areas between Little Porcupine Creek

and the Poplar River. Hills resulting from

erosion of beds of the Hell Creek and Fox Hills

ervation and in a small area in the southeast corner. Most beds easily erode on exposure and are distinguished by the darkness of the lignite. The Flaxville Formation, of Pliocene or Miocene age, consists of sand and gravel composed of siliceous pebbles such as quartzite, chert, opal, flint, chalcedony, and some silicified wood (Witkind, 1959). The gravels, which form the cap

of even-topped plateaus and benches, are at land

surface north of Cottonwood Creek and between

Little Porcupine Creek and the Poplar River. Pleistocene glacial deposits consist mainly of till, but also contain deposits of sand, gravel, and silt. These glacial deposits generally are present throughout the reservation, except for a gravel-capped plateau at the heads of Wolf Creek and Tule Creek drainages. The location of glacial deposits thicker than about 80 ft is shown in figure 3.

Holocene alluvium consists of flood-plain deposits of silt and local gravel lenses. These deposits are present along the Missouri River and major tributaries.

SUMMARY

Aquifers that are commonly used in the Fort Peck Indian Reservation for water supply exist in the Quaternary deposits (alluvium and glacial deposits), Flaxville Formation, Fort Union Formation, and a combination of the Fox Hills Sandstone and the lower part of the Hell Creek Formation. The alluvial aquifers are present mainly along the major stream valleys. Glacial aguifers are present in most of the reservation, but water yields differ greatly with location. The Flaxville aquifer is present in the northern and central parts of the reservation, and the Fort Union aquifer is present in the northcentral and eastern parts. The Fox Hills-lower Hell Creek aquifer is present in all areas of the reservation, except the extreme south-central and

The median specific conductance of representative samples of water collected from wells and springs in the reservation was 1,270 µS/cm, the median pH was 7.4, and the median temperature was 9.5 °C. Water from 10 wells completed in Quaternary aquifers and in the Madison Group and located in an area of water-quality concern north of Poplar had specific-conductance values greater than 4,500 µS/cm.

western parts.

Water samples from wells in the area of water-quality concern contained chloride concentrations that were much larger than from wells elsewhere in the reservation. Values of dissolved organic carbon also were larger in the area of water-quality concern than elsewhere.

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tana and North Dakota: U.S. Geological Survey Bulletin 1073, 80 p. CONVERSION FACTORS, ABBREVIATED WATER-QUALITY UNITS, AND VERTICAL DATUM

To obtain 4,047 square meter foot (ft) gallon per minute 0.06309 liter per second (gal/min) 25.4 millimeter inch (in.) mile (mi) 1.609 kilometer

square kilometer Temperature can be converted to degrees Fahrenheit (°F) or degrees Celsius (°C) by the equa-

 $^{\circ}F = 9/5 (^{\circ}C) + 32$

 $^{\circ}C = 5/9 \ (^{\circ}F - 32)$

2.59

square mile (mi²)

Abbreviated water-quality units used in report: μg/L micrograms per liter μS/cm microsiemens per centimeter at 25 °C

mg/L milligrams per liter Sea level: In this report "sea level" refers to the National Geodetic Vertical Datum of 1929 (NGVD of 1929) -- A geodetic datum derived from a general adjustment of the first-order level

nets of the United States and Canada, formerly called Sea Level Datum of 1929.

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Geologic Quadrangle Map GQ-67, scale

GROUND-WATER RESOURCES Aquifers

Principal aquifers in the reservation exist in Quaternary deposits, Flaxville Formation, Fort Union Formation, and a combination of the Fox Hills Sandstone and the lower part of the Hell Creek Formation. The Quaternary aquifers include alluvium and glacial deposits. Generalized areas of occurrence of the principal aquifers are shown in figures 5-8. The boundaries of the aquifers were interpreted from geology mapped by Colton (1955, 1963a-i, 1964) and Colton and Bateman

The alluvial aquifers are present mainly along major stream valleys (fig. 5). The aquifers are recharged by infiltration of precipitation and streamflow and by lateral flow from adjacent aquifers, and are discharged by streamflow, evapotranspiration, leakage to underlying aquifers, and water withdrawals. Rivers occupying the alluvial valleys affect the hydraulic gradient within the aquifer.

The glacial aquifers are present in most of the reservation. Because water yields from these aquifers differ greatly with location, their areal extent was not interpreted in this study. However, the location of wells completed in glacial aquifers is shown in figure 5. These aquifers are recharged by infiltration of precipitation and streamflow and are discharged by streamflow, evapotranspiration, springs, leakage to underlying aquifers, and water withdrawals. Hydraulic gradients in the glacial aquifers fluctuate according to local flow systems.

The Flaxville aquifer is present in the northern and central parts of the reservation (fig. 6) and is mostly unconfined. The aquifer is recharged by infiltration of precipitation and is discharged by evapotranspiration, springs, leakage to underlying aquifers, and water withdrawals. The flow system of the aguifer is affected by the many streams that cut through the aquifer. Springs are common in the lower part of

The Fort Union aquifer is present in the north-central and eastern parts of the reservation (fig. 7). The aquifer is recharged by infiltration of precipitation on the outcrops and by leakage from the overlying Flaxville and glacial aquifers and is discharged by subsurface outflow, evapotranspiration, and water withdrawals. Altitudes of measured water levels indicate a generally eastward flow of ground

the aquifer near stream channels.

The lower part of the Hell Creek Formation is hydraulically connected to the Fox Hills Formation. Together, these two units compose the Fox Hills-lower Hell Creek aquifer. This aquifer is present in all areas of the reservation except the extreme south-central and western parts (fig. 8). The aquifer is recharged by infiltration of precipitation on the outcrops and by leakage from overlying aguifers and is discharged by evapotranspiration at the outcrops and water withdrawals from the deeper and confined northeastern part of the aquifer. Altitudes of measured water levels indicate a generally eastward flow of ground water.

completed in the alluvial aquifers. The second most common source is wells completed in the Fox Hills-lower Hell Creek aguifer. East of 106° long., well yields in the reservation range from 5 to 25 gal/min for alluvial aquifers, 10 to 50 gal/min for glacial aquifers, 10 to 25 gal/min for the Flaxville aquifer, 2 to 10 gal/min for the Fort Union aquifer, and 10 to 25 gal/min for the Fox Hills-lower Hell Creek aquifer (Donovan, 1988, p. 34). The water withdrawn is used mostly for livestock watering and domestic purposes.

Most ground-water withdrawals are from wells

Secondary aquifers in the reservation are those deeper than the Fox Hills-lower Hell Creek aquifer. In this study, the deeper units inventoried were the Judith River aquifer of Late Cretaceous age and the Madison aquifer of Mississippian age. These secondary aquifers generally are not used for water sources in the reservation because of their depth and the thickness of the overlying Bearpaw Shale, which is difficult to

Water Levels and Physical Properties of the Water

At each site inventoried during 1989, water level, specific conductance, pH, and water temperature were measured where possible. Each aquifer has a distinct range in altitude of water levels and a distinct median specific conductance; therefore, each is discussed separately. Median values of specific conductance were calculated for water from the principal aquifers. To avoid overemphasizing multiple measurements, the median was computed using one value per well or spring; if multiple samples were available for a well or spring, the mean from the samples was used to represent the aquifer at that site. Except for well 32 completed in the Madison Group, values of pH and water temperature did not significantly differ from aquifer to aquifer, so only a median value for all water samples is

sentative samples of water collected from wells and springs in the reservation was 1,270 µS/cm. Water from 10 wells completed in Quaternary aquifers and the Madison Group in an area of water-quality concern north of Poplar (wells 32, 34-38, 56, 58, 59, 61) had specific-conductance values that were greater than 4,500 µS/cm. These samples were excluded from calculations of median specific conductance, because the water may be contaminated. Reservation-wide, the median pH was 7.4 and the median temperature was 9.5 °C.

The median specific conductance of repre-

During 1989, 35 wells completed in the alluvial aquifers were inventoried. Depth of wells ranged from 12 ft near the Poplar River (well 57) to 112 ft near Tule Creek (well 70). Altitudes of water level ranged from 1,895 ft near the Missouri River (well 40) to 2,743 ft near West Fork Wolf Creek (well 66). The median specific conductance was 1,480 µS/cm.

from glacial aquifers were inventoried during 1989. Depth of wells ranged from 13 ft in the northeastern part of the reservation (well 112) to 150 ft in the southeast (well 33). Altitudes of water level ranged from 1,924 ft in the southeast (well 16) to 2,857 ft in the northwest (well 79). The median specific conductance was 1,760 µS/cm.

Eighteen wells and 2 springs producing water

During 1989, 17 wells completed in the Flaxville aquifer were inventoried. Depth of wells ranged from 18 ft in the northwestern part of the reservation (well 78) to 100 ft in the central part (wells 48 and 49). Altitudes of water level ranged from 2,652 ft in the central part (well 88) to 2,899 ft in the northwest (well 106). The median specific conductance was 712 µS/cm.

Eighteen wells completed in the Fort Union aquifer were inventoried during 1989. Depth of wells ranged from 20 ft (well 117) to 310 ft (well 104) in the northeastern part of the reservation. Altitudes of water level ranged from 1,912 ft in the southeast (well 39) to 2,840 ft in the central part (well 85). The median specific conductance was 1,270 µS/cm.

During 1989, 27 wells completed in the Fox Hills-lower Hell Creek aquifer were inventoried. Depth of wells ranged from 51 ft in the northcentral part of the reservation (well 99) to 360 ft in the east-central part (well 72). Altitudes of water level ranged from 2,068 ft in the eastcentral part (well 72) to 2,875 ft in the north-central part (well 108). The median specific conductance was 940 µS/cm.

Two wells completed in deeper, secondary aquifers were inventoried during 1989--well 20 in the Judith River Formation, which is used for livestock watering, and well 32 in the Madison Group, which is used for oil production. Water levels were not measured in either well. The specific conductance was 4,200 µS/cm in well 20 and 200,000 μ S/cm in well 32.

Chemical Ouality

Of the 20 water samples analyzed during 1989, 12 are considered to be representative of the chemical quality of water in the principal aquifers in the reservation. However, only generalized statements can be made about the water quality of each aquifer owing to the small number of analyses. The remaining eight analyses are from wells in the area of water-quality concern north of Poplar (wells 32, 34, 37, 38, 56, 58, 59, 61); these analyses are described separately from the 12 representative analyses.

The ionic composition of the 20 water samples is depicted in a trilinear diagram (fig. 9) suggested by Piper (1944). Points on the diagram identify percentage composition rather than concentration. Therefore, two samples containing equal percentage distributions of calcium, magnesium, and sodium plus potassium would plot at the same point regardless of differences in actual concentrations.

Analyses of the water samples in table 2 are compared to primary and secondary drinking-water regulations established by the U.S. Environmental Protection Agency (1989a,b). Primary drinking-water regulations, which are mandatory for public water supplies, specify a maximum contaminant level (MCL) of 10 mg/L for nitrate (as nitrogen). Secondary drinking-water regulations, which apply to all public water supplies unless no better supply is available, include secondary maximum contaminant levels (SMCL) of 250 mg/L for sulfate, 500 mg/L for dissolved solids, 300 µg/L for iron, and 50 µg/L for manganese. The analyses of water samples in table 2 also are compared to a classification system for livestock watering developed by Montana State College (now Montana State University), Agricultural Experiment Station (McKee and Wolf, 1971, p. 113). The clas-

sification system is based on dissolved-solids

0 - 2,500 mg/L, good 2,500 - 3,500 mg/L, fair 3,500 - 4,500 mg/L, poor

Greater than 4,500 mg/L, unfit

concentration of the water:

Of the four representative water samples from alluvial aquifers, one (well 10) is dominated by the sodium and calcium cations and the sulfate anion; one (well 24) is dominated by the sodium cation and the sulfate anion; one (well 40) is dominated by the calcium cation and the bicarbonate and sulfate anions; and one (well 73) is dominated by the sodium cation and the bicarbonate anion (fig. 9). These water samples had concentrations (table 2) that greatly exceed the SMCL's for sulfate (wells 10, 24, 40) and dissolved solids (wells 10, 24, 40, 73). In addition, the concentration (16 mg/L) of the water sample from well 24 exceeded the MCL for nitrate, and the concentrations (table 3) of the water samples from wells 10 and 40 exceeded the SMCL's for iron and manganese. The suitability of water for livestock watering is good, on the basis of dissolved-solids concentration.

Of the two representative water samples from glacial aquifers, both (wells 7 and 21) are dominated by the sodium cation and the sulfate anion. Water samples from both wells had concentrations that exceed the SMCL's for sulfate and dissolved solids. Water from well 7 also had concentrations that exceed the SMCL's for iron and manganese. The suitability of water for livestock watering is good, on the basis of dissolvedsolids concentration.

Of the two representative water samples from the Flaxville aquifer, both (wells 69 and 97) are dominated by the calcium cation and the bicarbonate anion. Both water samples had concentrations that exceed the MCL for nitrate. The suitability of water for livestock watering is good, on the basis of dissolved-solids concen-

Of the two representative water samples from the Fort Union aquifer, one (well 86) is dominated by the magnesium cation and the bicarbonate anion, and one (well 104) is dominated by the sodium cation and the bicarbonate and sulfate anions. Water from well 104 had concentrations that exceed the SMCL's for sulfate and dissolved solids. The suitability of water for livestock watering is good, on the basis of dissolvedsolids concentration.

the Fox Hills-lower Hell Creek aquifer, both (wells 65 and 93) are dominated by the sodium cation and the bicarbonate anion. Water from well 93 had concentrations that exceed the SMCL's for sulfate, dissolved solids, and iron. The suitability of water for livestock watering is good, on the basis of dissolved-solids concen-

Of the eight water samples from wells in the

Of the two representative water samples from

area of water-quality concern north of Poplar, six (wells 32, 34, 56, 58, 59, 61) are dominated by the sodium cation and the chloride anion and two (wells 37 and 38) are dominated by the magnesium cation and the chloride anion; in contrast, water containing principally the sodium cation and the bicarbonate anion was sampled from well 73, which is upgradient from the eight sites and can be considered to be a control site. Percentage composition of chloride may be an indicator of water-quality contamination. Water samples from the eight wells in the area of water-quality concern contained percentage compositions of chloride greater than 70, whereas water samples from all other wells contained percentage compositions of chloride less than 25. Water samples from the eight wells had concentrations of chloride that ranged from 1,100 to 60,000 mg/L.

Analytical results for trace-element concentration and stable-isotope ratios for 20 water samples are given in table 3. These results provide background information for ground water in the study area. The detection limits for trace elements are affected by dissolved solids-generally the larger the dissolved-solids concentration of the water sample, the larger the

The results of analysis for dissolved organic carbon (DOC) also are given in table 3. In water samples from the eight wells having large percentage compositions of chloride in the area of water-quality concern, the concentrations of DOC were larger (median of 5.2 mg/L) than from other wells in the reservation (median of 2.6 mg/L), which indicates a source of DOC that may not be naturally occurring. Further investigation of the types of organic carbon dissolved in these waters could aid in determining the sources.

EXPLANATION

AQUIFER--Solid symbol is for representativ sample. Open symbol is for sample from area of water-quality concern. Numeral is

> Flaxville aquifer Fort Union aquifer Fox Hills - lower Hall Creek aquifer Madison Group

well number

▼ Alluvium

△ Glacial deposits

detection limit for the analytical method used.

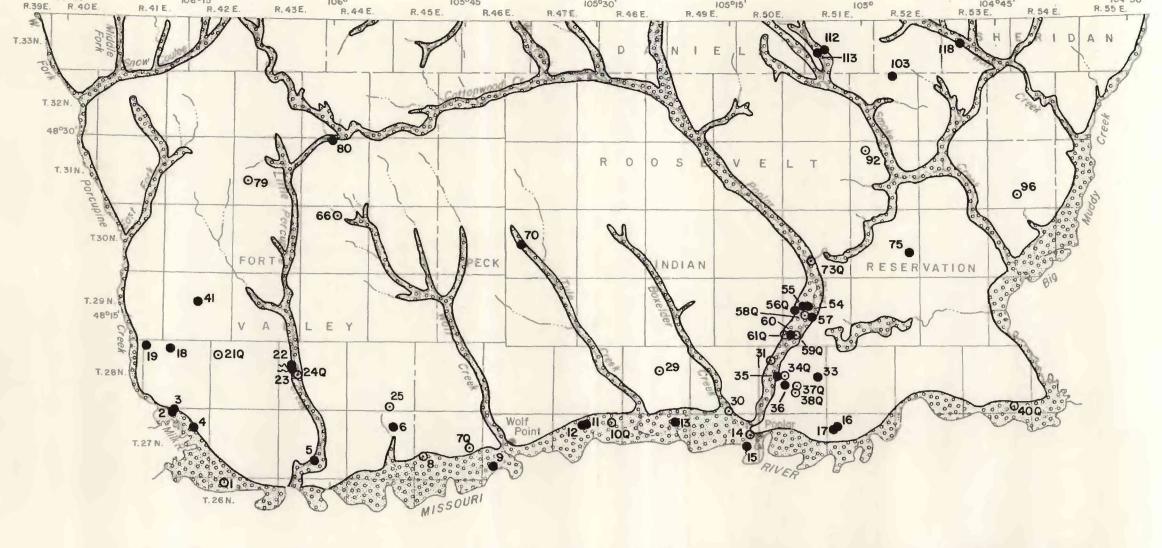


FIGURE 5.-GENERALIZED AREA OF THE ALLUVIAL AQUIFERS AND LOCATION OF WELLS AND SPRINGS COMPLETED IN QUATERNARY AQUIFERS.

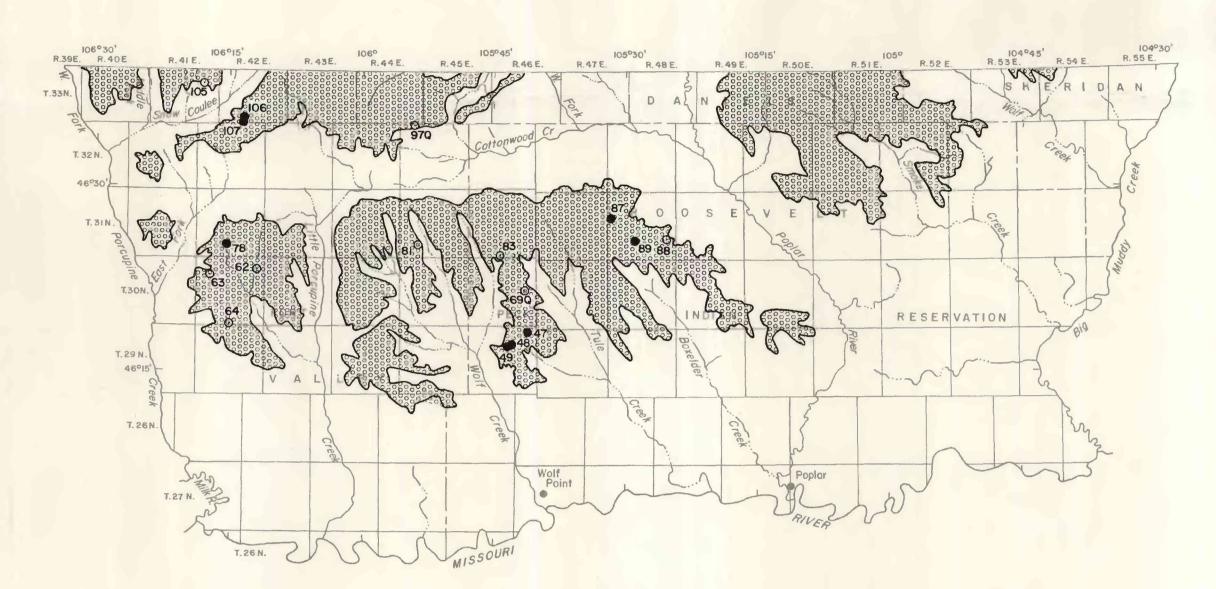


FIGURE 6.--GENERALIZED AREA OF THE FLAXVILLE AQUIFER AND LOCATION OF WELLS COMPLETED IN THE AQUIFER.

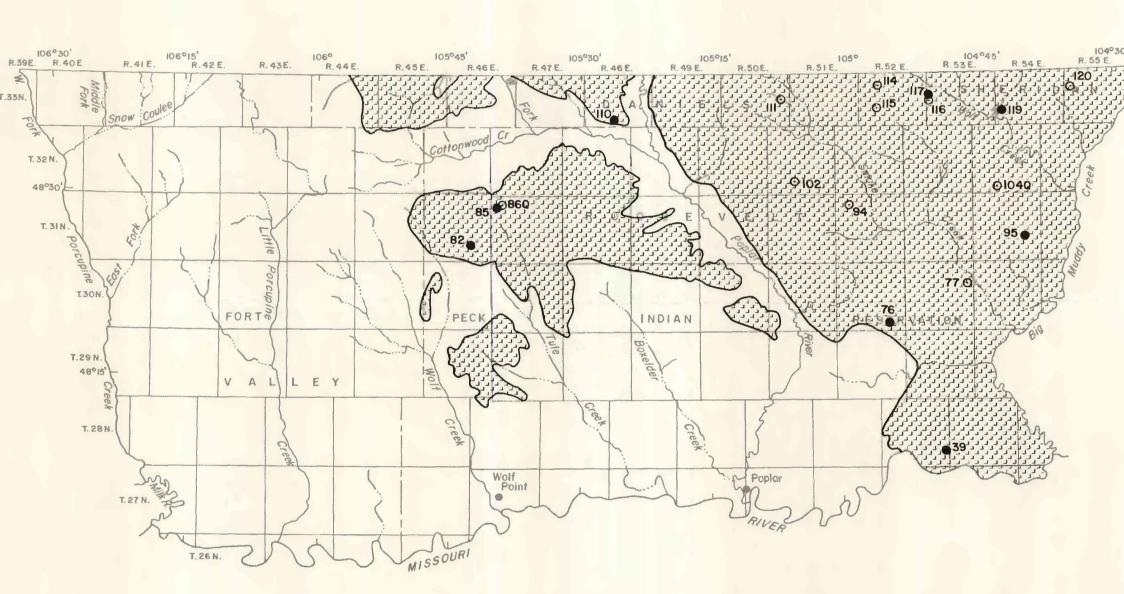


FIGURE 7.--GENERALIZED AREA OF THE FORT UNION AQUIFER AND LOCATION OF WELLS COMPLETED IN THE AQUIFER.

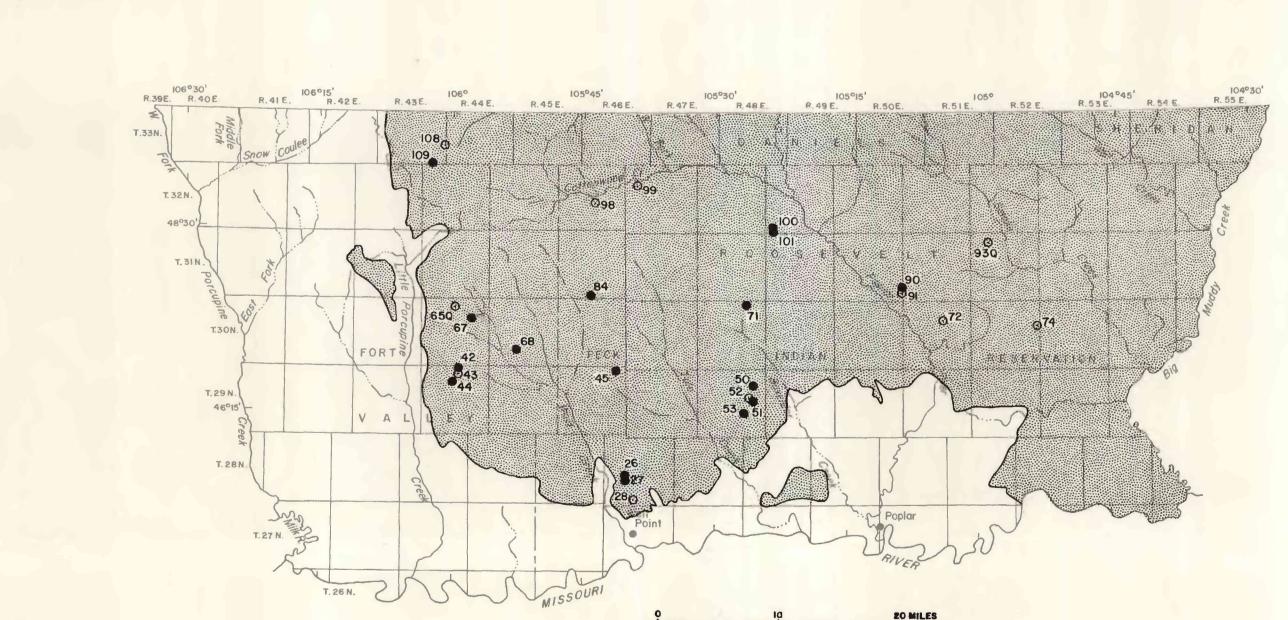


FIGURE 8 .- GENERALIZED AREA OF THE FOX HILLS-LOWER HELL CREEK AQUIFER AND LOCATION OF WELLS COMPLETED IN THE AQUIFER.

AREA OF ALLUVIAL AQUIFERS--Small areas not shown

EXPLANATION FOR FIGURES 5-8

- AREA OF FLAXVILLE AQUIFER AREA OF FORT UNION AQUIFER
- AREA OF FOX HILLS-LOWER HELL CREEK AQUIFER

- --- GENERALIZED CONTACT INVENTORIED WELL
- INVENTORIED WELL--Water level measured twice during 1989
- INVENTORIED SPRING
- NUMERAL IS WELL OR SPRING NUMBER Q WELL WITH WATER-QUALITY ANALYSIS

TABLE 1. -- RECORDS OF INVENTORIED WELLS AND SPRINGS

[-- or -, no data or not applicable]

Local number -- numbering system described in text. Type of site--C, collector well; S, spring; W, well. Altitude of land surface--in feet above sea level. Primary use of water--C, commercial; H, domestic; I, irrigation; S, livestock;

U, unused; Z, other.
Depth of well--in feet below land surface. Water level -- in feet below land surface.

Water-level status--P, pumping; R, recently pumped.
Water-level source--A, measured by Fort Peck Tribes; S, measured by U.S. Geological Altitude of water level -- in feet above sea level. Specific conductance--in microsiemens per centimeter at 25 degrees Celsius; onsite measurement.

pH--onsite measurement. Water temperature -- in degrees Celsius.
Principal aquifer -- Qal, Quaternary alluvium Og, Pleistocene glacial deposits

Tfg, Flaxville Formation
Tfu, Fort Union Formation Kfh-Khc, Fox Hills-lower Hell Creek aquifer Kjr, Judith River Formation Mm, Madison Group

Well or spring number	Local number	Type of site	Altitude of land surface (feet)	Primary use of water	Depth of well (feet)	Water level (feet)	Water- level status	level	Altitude of water level (feet)	Date of measure- ment	Specific con- ductance (µS/cm)	pH (units)	Water temper- ature (°C)	Principal aquifer	Well or spring number
1 2	26N43E05CAAB01 27N42E04ABBA01	W	2,035	н - U	38 94	16.36 15.86	:	s s	2,019 2,019	04-26-89 10-28-89	1,500	7.4	10.0	Qal Qal	1 2
3 4 5	27N42E04ABCC01 27N42E11CADC01 27N44E27BCCA01	W W	2,065 2,060 2,070	U н	90 100 43	9.39	=	s	2,051	04-26-89 06-08-89	3,400	7.3	11.0	Qal Qal	3 4 5
6 7	27N45E11CBBA01 27N46E23AAAB01	W W	2,210 2,060	н -	33 65	54.44 54.25	-	s s	2,006 2,006	06-09-89 06-09-89 07-21-89	910 3,390 3,200	7.4 7.2 7.3	10.5 10.5 12.0	Qg Qg	6 7
9	27N46E30AADB01 27N47E30AABD01	w - w	1,990	н - н	60 57	23.96	R - -	s s	1,996	10-24-89 04-27-89	1,490 1,180 1,310	7.7 7.2 7.7	9.0 11.0 9.5	Qal Qal	9
10	27N48E01CCCD01 27N48E04DACB01	W - W	1,980	H	32 25	19.00 22.75	P -	s s -	1,961 1,957	06-10-89 07-20-89 	2,200 2,650 	7.9 7.1	11.5	Qal Qal	10
12 13 14	27N48E04DACC01 27N49E02CAAD01 27N50E12CCBC01	W W	1,980 1,970 1,955	н н н	38 72 60	20.12 17.23 14.00 14.20	- - P	S S S	1,960 1,953 1,941 1,941	04-28-89 06-12-89 06-10-89 08-03-89	2,720 1,420 1,800 1,850	7.5 8.5 7.7 7.4	8.0 10.0 13.0 14.0	Qal Qal Qal	12 13 14
15 16 17	27N50E14DDAA01 27N52E07DABB01 27N52E07DABB02	W W	1,960 1,995 1,995	H U	125 90	71.44	-	s	1,924	06-10-89 06-12-89	825 2,700	7.9	8.0	Qal Qg Qg	15 16 17
18 19 20	28N42E04CBAA01 28N42E06BACD01 28N42E33CDDD01	W W	2,405 2,275 2,100	H H S	22 17 635	12.07 4.25	- :	s s	2,393 2,271	04-27-89 05-14-89 04-26-89	2,810 1,500 4,200	7.3 7.3 8.1	6.5 6.0 8.0	Qal Qal Kjr	18 19 20
21 22 23	28N43E07AABD01 28N44E17CCBD01 28N44E17CCCD01	W - S S	2,440 2,345 2,340	S S H	64 	56.18 56.70	P -	s s	2,384 2,383	06-07-89 07-21-89 06-08-89 06-08-89	1,520 1,500 2,000 1,440	7.5 7.4 7.7 7.7	9.0 11.0 9.0 9.5	Qg Qg Qg	21 22 23
24	28N44E17CDCC01 28N45E34ADDC01	W - W	2,280	H ~ H	40 56	9.68 10.56 44.86	:	s s	2,270 2,269 2,260	06-08-89 07-21-89 06-09-89	2,510 3,000 1,100	7.3 7.2 7.6	8.5 9.5	Qāl Qg	24 25
26 27	28N47E20DADB01 28N47E20DADB02	w w	2,355 2,355	H C	137 137	45.15 		s - -	2,260 	08-04-89 05-01-89 05-01-89	1,600 1,090 940	7.1 8.0 7.8	9.0 9.0 14.0	Kfh-Khc Kfh-Khc	26 27
28 29	28N47E33CBCA01 28N49E15BACA01	w - -	2,225 2,110	S H -	110	43.78 44.80 8.89 10.67	- - P	S S S	2,181 2,180 2,101 2,099	05-01-89 08-04-89 06-10-89 10-30-89	1,300 1,300 3,900 2,600	7.9 7.1 8.6 7.3	9.0 10.0 13.0 16.0	Kfh-Khc Qal	28 29
30	26N50E34CDBC01 26N51E08BBDB01	w - w	1,970 1,995	н - s	55 14	24.73 25.05 6.75	R - -	s s	1,945 1,945 1,988	04-19-89 08-03-89 06-10-89	2,100 1,850 2,120	7.2 7.1	9.0 9.5	Qal Qal	30 31
32 33	28N51E10DBAD01 28N51E13CCCC01	CW	2,180	Z H	150	11.04	:	s s	1,984 2,047	10-25-89 07-24-89 05-16-89	2,900 200,000 2,990	7.0 6.4 7.2	10.0 80.0 11.0	Mm Qg	32 33
34 35 36	28N51E16CACC01 28N51E17DDAD01 28N51E21BCAD01	W - W	2,040 2,025 2,010	U - н н	104	79.72 79.60 	=	s - -	1,960 1,960 	07-23-89 10-25-89 07-22-89 07-20-89	10,000 5,200 6,500	7.3	10.0	Qg Qg Qg	34 35 36
37	26N51E22CDCC01 26N51E27BBBB01	- W	2,105	н - z	85 68	59.99 32.50	:	s s	2,045 2,045 2,053	07-20-89 10-26-89 07-24-89	13,000 11,400 25,000	6.8 6.7 6.8	11.0 10.5 10.5	Qg Qg	37 38
39 40	28N53E25DBBC01 28N54E35CAAB01	w w	1,950 1,920	н	82 45	42.43 37.56 23.86	=	s s	2,043 1,912 1,896	10-25-89 05-13-89 06-12-89	22,500 2,520 1,380	6.3 7.3 7.3	10.0 9.5 9.5	Tfu Qal	39 40
41 42 43	29N42E15CAAB01 29N44E03BBCA01 29N44E03BBCD01	W W	2,650 2,730 2,720	H H S	16 120 90	24.90 51.22 56.78	P - - P	s s s	1,895 2,679 2,663	07-20-69 10-24-69 05-15-69	1,420 560 560	7.2 7.3 7.7	9.0 8.0	Qg Kfh-Khc Kfh-Khc	41 42 43
44 45	29N44E09ACDD01 29N46E01BCDB01	w w	2,740 2,700	- н s	130 270	55.20 69.58	P -	s s	2,664	10-24-89 05-15-89 05-01-89	560 760 730	7.5 7.7 7.9	9.5 9.5 9.5	Kfh-Khc Kfh-Khc	44 45
46 47 48	29N46E01BCDB02 29N46E01BCDB03 29N46E10DAAB01	W W	2,700 2,700 2,705	U S	45 100	34.81	Ξ	s	2,665	05-01-89 05-01-89 05-01-89	530 510	7.4	15.0	Tfg Tfg	46 47 48
49 50 51 52	29N46E10DAAC01 29N48E11ADDD01 29N48E14DBCC01 29N48E14DCAB01	WWW	2,705 2,490 2,400 2,400	H K S H	100 140 65 65	34.02	Ē	- - s	2,366	05-01-89 04-29-89 04-28-89	685 520 495	7.9 7.8 7.4	11.0 10.0 8.5	Tfg Kfh-Khc Kfh-Khc Kfh-Khc	49 50 51 52
53 54	29N48E22DDDD01 29N51E16CBCC01	W	2,440	H	190 25	34.78 15.22		s - s	2,365	10-30-89 07-22-89 06-11-89	1,200	6.9	9.5	Kfh~Khc	53 54 55
55 56 57 58	29N51E16CCAA01 29N51E17CCCC01 29N51E21ABCC01 29N51E21BBAA01	W W W	2,025 2,065 2,015 2,025	U H U	23 87 12	11.85 56.90 11.47		s s s	2,013 2,008 2,004	07-22-89 06-11-89 04-19-89	9,500 1,500	8.0 7.3 7.7	10.5	Qal Qal Qal	56 57 58
59 60	29N51E31AABB01 29N51E32BBBA01	W - W	2,009	- U	54	11.85 20.48 20.75 8.81	-	s s s	2,013 1,989 1,988 1,991	07-22-89 04-19-89 07-23-89 05-02-89	4,600 12,000 13,500	7.7 6.8 7.1	8.5 14.0 10.0	Qal Qal	59 60
61	29N51E32BBBB01 30N42E01CDBC01	W - W	2,000	U ~	49	10.20 9.60 12.11	1	s s s	1,990 1,990 2,728	04-19-89 07-23-89 06-07-89	30,000 65,000 5,060	6.8 6.7 7.4	16.0 11.0	Qal Tfg	61 62
63	30N42E08BBAA01	w	2,730	н -	50	13.30 25.92 26.08		s s	2,727 2,704 2,704	08-04-89 06-07-89 10-28-89	4,800 1,540	6.8 7.3	10.5	Tfg	63
64 65	30N42E33DDDD01 30N44E03CBBB01	W	2,755 2,835	H - S	60 170	46.82 47.60 97.69 96.08	- - R	s s s	2,708 2,707 2,737 2,739	05-14-89 08-04-89 05-15-89 07-21-89	2,240 2,000 370 430	7.5 6.9 8.2 7.8	9.5 9.0 9.0 8.5	Tfg Kfh-Khc	64 65
66 67	30N44E03CDAC01 30N44E11CCAB01	W - W	2,770	s - I	105 275	34.57 37.14 85.93		s s	2,735 2,743 2,699	05-15-89 10-24-89 05-15-89	705 660 1,130	7.3 7.1 8.2	8.5 8.5 10.0	Qal Kfh-Khc	66 67
68 69	30N45E28BBDB01 30N46E14DDDD01	W	2,735 2,715	H H	176	9.41 9.94	R -	s s	2,581 2,706 2,705	06-08-89 05-01-89	555 555 530	6.9 7.7 7.5	9.0 11.0	Kfh-Khc Tfg	68 69
70 71 72	30N47E20BBBB01 30N48E02BBDA01 30N51E10CCAD01	W W -	2,545 2,635 2,165	S H H	112 265 360	16.33 96.03 96.61		s - s s	2,529 2,069 2,068	05-02-89 04-29-89 05-12-89 10-25-89	1,020 1,550 1,750	7.3 7.4 8.6 8.8	9.0 7.5 10.5 11.0	Qal Kfh-Khc Kfh-Khc	70 71 72
73 74	30N51E28DCAA01 30N52E13ABAB01	W - W	2,060	H I	20 276	11.05 11.15 48.46 75.03	Ē	S S S	2,049 2,049 2,182 2,155	07-22-89 10-25-89 05-11-89 10-29-89	1,340 1,150 2,520 2,350	7.4 7.8 8.8 8.5	9.5 10.0 8.5 9.5	Qal Kfh-Khc	73 74
75 76 77	30N52E24CCCC01 30N52E35AAAA01 30N53E12DCCA01	W	2,160 2,135 2,065	н н	60 120 120	45.81		- - s	2,019	 06-12-89 05-11-89	1,650 1,530	8.3 6.8	10.5 9.5	Qg Tfu Tfu	75 76 77
78 79	31N42E28DCDD01 31N43E20CDDC01	W	2,785 2,880	— н	16 30	46.40 10.28 22.95	P - -	s s	2,019 2,775 2,857	10-29-89 06-07-89 06-07-89	1,550 4,030 940	6.7 7.2 7.2	9.5 12.0 12.0	Tfg Qg	78 79
80 81	31N44E04AAAC01 31N45E29DDCA01	w w	2,670 2,835	н н	25 50	24.95 13.85 23.22 26.90	R -	SSSS	2,855 2,656 2,812 2,808	08-04-89 05-15-89 06-08-89 10-24-89	910 830 920 920	7.2 7.4 7.6 7.5	13.0 9.5 8.5 9.5	Qal Tfg	80 81
82 83	31N46E26BCCC01 31N46E33DDAA01	w -	2,730	н -	60	15.04 49.60 54.48	- - P	S S A	2,715 2,675 2,671	05-01-89 05-01-89 08-22-89	580 695 560	7.3 7.6 6.3	7.0	Tfu Tfg 	82 83
84 85 86	31N46E33DDAB01 31N47E06DDDD01 31N47E06DDDD02	W W	2,725 2,870 2,870	z s H	55 60	30.10 34.99	=	s s	2,840 2,835	06-13-89 05-02-89 05-02-89	470 655 465	7.5 7.8 7.4	7.5 7.0 9.5	Kfh-Khc Tfu Tfu	84 85 86
87 88	31N48E18BCBC01 31N48E25BCCA01	W	2,740 2,680	н н –	30 30	38.91 27.82 27.87	Ē	s s s	2,831 2,652 2,652	07-19-89 04-29-89 04-29-89 08-04-89	440 650 2,350 1,900	7.5 7.6 7.5 7.0	10.0 10.0 8.0 8.0	Tfg Tfg	87 88
89 90 91	31N48E28BABD01 31N50E25DDDD01 31N50E36AAAA01	W W	2,710 2,220 2,215	н н	40 185 150	99.93	=	- s	2,115	04~29-89 06-13-89	1,900	7.3 8.7	12.5	Tfg Kfh-Khc Kfh-Khc	90 91
92	31N52E05CDCD01	W	2,395	- s -	48	100.32 15.45 17.11	-	s s	2,115 2,380 2,378	10-25-89 06-12-89 10-29-89	1,900 1,180 1,160	8.8 7.6 6.5	9.5 7.0 8.0	Qal	92
93 94	31N52E05CDDB01 31N52E05CDDB02	W	2,400	H - Z -	330 60 	43.61 45.08 34.24 35.66	R	s s s	2,356 2,355 2,371 2,369	06-12-89 07-18-89 06-12-89 10-29-89	3,000 2,150 1,060 1,040	8.6 8.4 7.6 7.1	10.0 11.0 9.0 9.5	Kfh-Khc Tfu	93 94
95 96	31N54E23DDDD01 31N54E28DCDC01	W - 3	2,095	H S ~	80 50	12.21 14.60	=	s	2,068	05-11-89 05-11-89 10-29-89 05-15-89	1,230 2,520 2,400	7.1 7.1 6.8	9.5 8.0	Tfu Qg	95 96
97 98	32N45E05BBCC01 32N46E22BBDA01	w w	2,890	н - s	110	29.14 34.05 32.04	R R	s s s	2,861 2,856 2,413	05-15-89 07-19-89 05-01-89 10-28-89	535 500 1,390	7.3 7.4 8.9	8.0 9.5 9.5	Tfg Kfh-Khc	97 98
99 100	32N47E07DAAD01 32N49E31DCBB01	- - - -	2,400	H S H	51 183 200	41.74 41.44	=	s s	2,417 2,358 2,359 	10-28-89 05-02-89 10-28-89 04-29-89	1,320 1,340 1,460	8.9 7.5 7.1	8.0 9.5 9.0	Kfh-Khc	99 100
101 102 103	32N49E31DCDC01 32N51E28DCCD01 32N52E03AADC01	W - W	2,390 2,655 2,560	н - н	200 249 65	201.19 153.13 44.78	P -	s s	2,454 2,502 2,515	04-29-89 05-12-89 10-27-89 05-10-89	1,190 1,400 1,380 810	9.1 6.9 7.0 7.0	9.5 9.5 9.0	Kfh-Khc Tfu Qal	101 102 103
103	32N54E33BABC01 32N54E33BABC01 33N41E14BADD01	W - W	2,355	н 2 -	310 75	161.50 165.04	=	s s s	2,515 2,194 2,190 2,831	05-10-89 05-11-89 07-18-89	2,420 2,100 515	7.0 7.5 7.7 7.6	10.0 10.5	Qal Tfu Tfg	103
106 107 108	33N42E33BCCC01 33N42E33CCBB01 33N43E25ADAD01	W W	2,930 2,905 2,905	H I H	95 75 165	54.10 30.65 8.42 30.00	=	s s s	2,831 2,899 2,897 2,875	08-04-89 06-07-89 06-07-89 04-18-89	445 740 970 440	6.9 7.9 6.9 7.3	8.0 10.0 6.0 10.0	Tfg Tfg Kfh-Khc	106 107 108
109 110	33N43E35DDDD02 33N48E33ADAD01	- W W	2,930 2,350	- н н	165 37	29.59 71.82 20.68	- R	s s	2,875 2,858 2,329	10-24-89 04-18-89 04-29-89	440 570 2,490	7.4 7.3 7.9	8.5 10.0 9.0	Kfh-Khc	109 110
111	33N50E24BDDC01 33N51E20DCCC01	W - W	2,540	H - H	13	9.59 9.41 10.50	=	s s	2,530 2,531 2,490	05-12-89 10-27-89 05-12-89	1,310 1,320 1,130	7.2 7.1	9.0 9.0 14.0	Tfu Qg	111
113 114 115	33N51E29BBBA01 33N52E17AAAB01 33N52E29AAAA01	W W W	2,505 2,410 2,645	н - н	38 50 168	15.43 27.99 28.25 140.69	P -	s s s	2,490 2,382 2,382 2,504	05-12-89 05-11-89 10-27-89 05-10-89	2,150 890 960 730	6.9 7.0 7.1 7.2	11.0 9.5 10.0 8.5	Qg Tfu Tfu	113 114 115
116 117	33N53E19CCBB01 33N53E19CCBC01	- W - W	2,370	- Н - s	54 20	138.25 34.70 35.42	=	s s	2,507 2,335 2,335	10-27-89 05-10-89 10-27-89 05-10-89	590 1,290 1,080 1,230	6.8 6.7 6.7 7.3	8.5 11.5 10.0 9.5	Tfu Tfu	116 117
118 119 120	33N53E20ACAB01 33N54E30DBCD01 33N55E18DDAA01	W W	2,210 2,170 2,135	H H H	35 100 200	13.16 38.20	=	s - s	2,197 2,097	05-10-89 05-10-89 05-10-89	1,950 2,400 2,240	7.5 7.4 6.9	9.0 8.0	Qal Tfu Tfu	118 119 120
		-		-		44.15		Α	2,091	08-29-89	1,600	7.8			

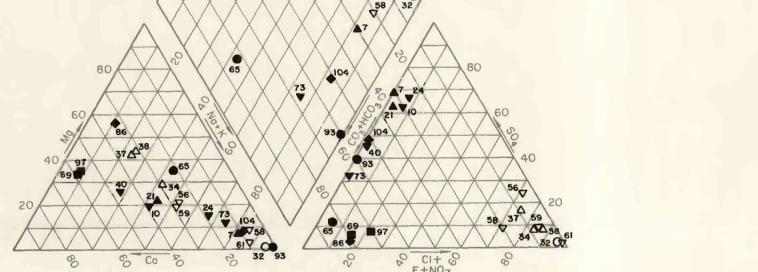


FIGURE 9 .-- TRILINEAR DIAGRAM SHOWING CHEMICAL-QUALITY DATA. DATA ARE CONVERTED FROM MILLIGRAMS PER LITER TO MILLIEQUI-VALENTS PER LITER, THEN VALUES FOR EACH CATION AND AN-ION ARE PLOTTED AS A PERCENTAGE OF TOTAL MILLI-EQUIVALENTS PER LITER.

CATIONS PERCENTAGE REACTING VALUES

TABLE 2 .-- PHYSICAL PROPERTIES AND MAJOR-ION CONCENTRATIONS IN WATER FROM SELECTED WELLS

[Principal aquifer: defined in table 1. Depth of well: in feet below land surface. Abbreviations: mg/L, milligrams per liter; fet, fixed-endpoint titration; fld, field (onsite) measurement; lab, laboratory. Symbols: <, less than detection limit for analytical method used; --, no data

Well number	Princi- pal aquifer	Date	Depth of well (feet)	Dis- solved oxygen (mg/L)	Hard- ness (mg/L as CaCO ₃)	Hard- ness, noncar- bonate (mg/L CaCO ₃)	Cal- cium, dis- solved (mg/L as Ca)	Magne- sium, dis- solved (mg/L as Mg)	Sodium, dis- solved (mg/L as Na)	Sodium ad- sorp- tion ratio	Potas- sium, dis- solved (mg/L as K)	Bicar- bonate, fet-fld (mg/L as HCO ₃)	Car- bonate, fet-fld (mg/L as CO ₃)	Alka- linity, field (mg/L as CaCO ₃)	Alka- linity, lab (mg/L as CaCO ₃)	Sulfate, dis- solved (mg/L as SO ₄)	Chlo- ride, dis- solved (mg/L as Cl)	Fluo- ride, dis- solved (mg/L as F)	Bromide, dis- solved (mg/L as Br)	Iodide, dis- solved (mg/L as I)	Silica, dis- solved (mg/L as SiO ₂)	Solids, sum of constit- uents, dis- solved (mg/L)	Nitro- gen, NO2+NO3 dis- solved (mg/L as N)	Well number
7 10 21 24 32	Qg Qal Qg Qal Mm	07-21-89 07-20-89 07-21-89 07-21-89 07-24-89	32 64 40	0.0 .1 9.0 7.1	280 680 460 510 3,000	0 300 180 100 2,800	67 180 110 120 950	28 55 45 51 140	630 220 170 480 39,000	16 4 3 9 310	4.5 9.1 4.6 5.8	653 473 353 514	0 0 0 0	526 379 285 411	542 379 284 425 183	1,200 730 530 1,000 1,800	14 62 16 56 60,000	0.6 .4 .2 .4 5.6	0.10 .10 .16 .29	0.013 .014 .007 .014 .880	14 18 16 12 86	2,280 1,510 1,060 1,970 103,000	0.12 .16 2.3 16 <.10	7 10 21 24 32
34 37 38 40 56	Qg Qg Qg Qal Qal	07-23-89 07-20-89 07-24-89 07-20-89 07-22-89	85 68 45	.0 .2 .0 .0	2,900 5,600 11,000 550 2,600	2,600 4,900 11,000 150 2,300	580 920 1,800 140 600	360 790 1,600 48 270	940 870 1,700 99 1,300	8 5 7 2 11	16 15 28 7.2	375 870 622 492 425	0 0 0 0 0	302 698 502 399 345	256 707 488 397 304	490 1,400 1,200 360 1,400	3,100 4,100 9,700 5.6 3,000	.1 .2 <.1 .6	.77 3.5 5.8 .03 3.3	.049 .560 .066 .007	20 29 36 24 17	5,720 8,570 16,400 931 6,840	<.10 15 2.8 .16 <.10	34 37 38 40 56
58 59 61 65	Qal Qal Qal Kfh-Khc Tfg	07-22-89 07-23-89 07-23-89 07-21-89 07-19-89	54 49 170	.0 .0 5.9 7.0	300 3,400 5,000 120 270	3,100 4,700 0	55 790 1,500 19 67	40 350 290 18 25	820 1,800 17,000 44 11	21 13 110 2	9.3 15 83 1.6 2.1	539 408 394 259 279	0 0 0	440 331 310 209 227	435 328 262 194 231	170 740 770 30 15	1,100 4,800 30,000 5.8 5.9	.3 .1 .1 .4	1.2 3.7 9.2 .07	.110 .150 1.1 .011 .003	15 20 19 12 14	2,480 8,760 49,900 257 277	<.10 .41 <.10 .64	58 59 61 65 69
73 86 93 97	Qal Tfu Kfh-Khc Tfg Tfu	07-22-89 07-19-89 07-18-89 07-19-89 07-18-89	330 50	9.5 -0 9.0	180 210 20 250 180	0 20 0 66	35 30 4.5 62 32	22 32 2.2 24 25	230 10 710 9.9 450	8 .3 69 .3	4.6 2.5 2.7 2.8 5.6	586 227 1,180 233 803	0 0 24 0	470 187 993 188 652	493 191 1,010 192 647	220 8.0 610 19 590	9.5 7.1 20 3.4 3.5	1.9	.07 .02 .24 .04	.011 .002 .041 .002	12 18 7.5 15	817 221 1,960 250 1,510	<.10 7.8 .85 16 <.10	73 86 93 97

TABLE 3.--TRACE-ELEMENT AND ORGANIC-CARBON CONCENTRATIONS AND STABLE-ISOTOPE RATIOS IN WATER FROM SELECTED WELLS

[Principal aquifer: defined in table 1. Depth of well: in feet below land surface. Abbreviation: µg/L, micrograms

Well number	Princi- pal aquifer	Date	Depth of well (feet)	Barium, dis- solved (µg/L as Ba)	Beryl- lium, dis- solved (µg/L as Be)	Boron, dis- solved (µg/L as B)	Cadmium, dis- solved (µg/L as Cd)	Chro- mium, dis- solved (µg/L as Cr)	Cobalt, dis- solved (µg/L as Co)	Copper, dis- solved (µg/L as Cu)	Iron, dis- solved (µg/L as Fe)	Lead, dis- solved (µg/L as Pb)	Lithium, dis- solved (µg/L as Li)	Manga- nese, dis- solved (µg/L as Mn)	Molyb- denum, dis- solved (µg/L as Mo)	Nickel, dis- solved (µg/L as Ni)	Silver, dis- solved (µg/L as Ag)	Stron- tium, dis- solved (µg/L as Sr)	Vana- dium, dis- solved (µg/L as V)	Zinc, dis- solved (µg/L as Zn)	Carbon, organic dis- solved (mg/L as C)	H-2/H-1 stable isotope ratio, per mil	0-18/0-16 stable isotope ratio, per mil	Well number
7 10 21 24 32	Qg Qal Qg Qal Mm	07-21-89 07-20-89 07-21-89 07-21-89 07-24-89	65 32 64 40	6 29 13 13 590	<1 <.5 <.5 <1 <50	1,200 160 270 480 20,000	<2.0 <1.0 <1.0 3.0	<10 <5 <5 <10 <500	<6 <3 <3 <6 <300	<20 <10 <10 <20 <1,000	1,300 3,000 18 10 690	<20 <10 <10 <20 <1,000	300 96 80 250 5,200	370 3,400 <1 <2 240	<20 <10 <10 <20 <1,000	<20 <10 <10 <20 <1,000	3.0 <1.0 <1.0 3.0 <100	730 1,500 760 1,000 54,000	<12 <6 <6 <12 <600	14 19 24 18 630	1.6 2.8 1.0 6.3 7.0	 -138.5	 -16.25	7 10 21 24 32
34 37 38 40 56	Qg Qg Qg Qal Qal	07-23-89 07-20-89 07-24-89 07-20-89 07-22-89	104 85 68 45 87	140 52 78 30 15	<3 <5 <10 <.5 <5	680 660 670 170 500	7.0 <10 <20 <1.0 <10	<30 <50 <100 <5 <50	<20 <6 <60 <3 <30	<50 <100 <200 <10 <100	24,000 430 1,100 5,400 26,000	<50 <100 <200 <10 <100	320 560 880 75 480	3,300 6,600 940 400 1,500	<50 <100 <200 <10 <100	<50 <100 <200 <10 <100	<5.0 <10 <20 <1.0 <10	5,500 13,000 29,000 1,300 5,500	<30 <60 <120 <6 <60	62 100 260 4 <30	1.4 5.5 6.5 2.9 4.5	-139.5 -141.5 -135.0 -135.5	-17.40 -18.35 -17.70 -17.00	34 37 38 40 56
58 59 61 65	Qal Qal Qal Kfh-Khc Tfg	07-22-89 07-23-89 07-23-89 07-21-89 07-19-89	17 54 49 170 40	89 67 150 88 590	<2 <5 <25 <.5 <.5	1,600 730 5,500 90 50	4.0 <10 <50 <1.0 <1.0	<20 <50 <250 <5 <5	<9 <30 <150 <3 <3	<30 <100 <500 <10	5,400 27,000 21,000 12 13	<30 <100 <500 <10 <10	190 550 2,300 25 14	230 2,000 6,400 <1 <1	<30 <100 <500 <10 <10	<30 <100 <500 <10 <10	6.0 17 160 1.0 <1.0	710 12,000 52,000 210 390	<18 <60 <300 <6 <6	<9 66 <150 4 33	3.0 5.0 7.0 .2 2.4	-138.5 -134.0 -122.0	-17.05 -16.80 -16.15	58 59 61 65
73 86 93 97	Qal Tfu Kfh-Khc Tfg Tfu	07-22-89 07-19-89 07-18-89 07-19-89 07-18-89	20 60 330 50 310	53 91 9 280 8	<.5 <.5 <1 <.5 <1	760 40 830 40 490	<1.0 <1.0 <2.0 <1.0 <20	<5 <5 <10 <5 <10	<3 <3 <6 <3 <6	10 10 <20 <10 <20	74 7 490 7 <12	<10 <10 <20 <10 <20	72 15 71 14 75	17 <1 8 <1 35	<10 <10 <20 <10 <20	<10 <10 <20 10 <20	1.0 1.0 <2.0 <1.0 3.0	370 380 160 440 1,200	<6 <6 <12 <6 <12	6 100 <6 19	3.2 1.2 5.5 1.7 5.0	-122.0 	-14.70 	73 86 93 97 104